

# Selective Catalytic/Non-Catalytic Reduction Conference

## Ammonia Slip and Increased SO<sub>3</sub> from SCR: Balancing Air Heater Deposits, Ammonia in Effluent Discharge, and SO<sub>3</sub> Plume

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May 16, 2002

# Agenda

- Introduction
- Formation and fate of ammonia
- Formation and fate of  $\text{SO}_3$
- Air heater deposits
- Ammonia in effluent discharge
- $\text{SO}_3$  mitigation
- Balancing ammonia slip and  $\text{SO}_3$
- Conclusions
- Recommendations

# Introduction

- SCR produces
  - Ammonia ( $\text{NH}_3$ ) slip
  - Increased sulfur trioxide ( $\text{SO}_3$ )
- Impacts on coal-fired power plants
  - High acid dew point temperature
  - Increased air heater deposits
  - $\text{NH}_3$  in effluent discharge
  - Potential for visible (“blue”)  $\text{SO}_3$  plume

# Formation and Fate of Ammonia

## ■ Formation

- Catalyst life – up to 20,000 hours
- <1 ppm  $\text{NH}_3$  for first half of catalyst life
- ~2 ppm  $\text{NH}_3$  at end of catalyst life

## ■ Fate – removal of $\text{NH}_3$ slip

- Air heater deposits – reaction of  $\text{NH}_3$  and  $\text{SO}_3$  to form ammonia bisulfate (ABS)
- Ad/absorbed on fly ash
- Absorbed in flue gas desulfurization (FGD) slurry

## ■ Impact

- Increased air heater deposits – ABS and fly ash
- $\text{NH}_3$  in fly ash/FGD effluent discharge

## Formation and Fate of Ammonia (Cont' d)

- Few reports on distribution of  $\text{NH}_3$  slip
  - Some of air heater deposits recycled back to boiler as  $\text{NO}_x$  - Ljunstroms
  - Best estimate - 50 to 80 percent on fly ash

# Formation and Fate of SO<sub>3</sub>

## ■ Formation

- Boiler SO<sub>3</sub> – 0.5 percent to 1.5 percent of SO<sub>2</sub>
- SCR SO<sub>3</sub> - ~1 percent of SO<sub>2</sub> (SCR ~doubles SO<sub>3</sub>)

## ■ Fate – removal of SO<sub>3</sub> (“inherent removal”)

- Condensate on air heater surfaces
- Air heater ash and ABS deposits
- Electrostatic precipitator (ESP) fly ash
- FGD slurry

## ■ Impact of increased SO<sub>3</sub>

- Higher acid dew point temperature
- Increased air heater deposits – fly ash and ABS
- Potential for visible plume

# Formation and Fate of SO<sub>3</sub> (cont' d)

## ■ Inherent SO<sub>3</sub> Removal

- With increased SO<sub>3</sub> from SCR, does percent removal or total ppm removed remain constant ?
- Boiler SO<sub>3</sub>: 20 ppm with 50% “inherent” removal = 10 ppm at stack
- Boiler + SCR SO<sub>3</sub> = 20 + 20 = 40 ppm
  - with 50% “inherent” removal = 20 ppm at stack
  - with “constant” 10 ppm removed = 30 ppm at stack

# Air Heater Deposits

## ■ General

- Ammonia bisulfate (ABS) is primary product from reaction of  $\text{NH}_3$  and  $\text{SO}_3$
- ABS condenses (initial formation temperature) at  $\sim 350^\circ\text{F}$  and is “sticky”
- IFT is a function of  $\text{SO}_3$  and  $\text{NH}_3$  – higher  $\text{SO}_3$  and  $\text{NH}_3 \rightarrow$  higher IFT

## ■ Thermodynamic and kinetic analysis

- High  $\text{SO}_3 : \text{NH}_3$  ratio  $\rightarrow$  high driving force for reaction and shorter temperature drop for completion of reaction
- Turbulent gas flow results in condensation of ABS on cooler heat transfer surfaces
- Laminar gas flow may produce ABS aerosol formation, which may pass through air heater (Results from rapid gas cooling before ABS diffuses to cooler air heater surfaces and condenses)



# Air Heater Deposits (Cont' d)

## ■ Radian Equation

- $\text{SO}_3 \text{ ppm} \times \text{NH}_3 \text{ ppm} \times [T (\text{IFT}) - T (\text{Rep})]$   
where “T(Rep)” is a combination of exit gas temperature and cold end metal temperature
- For 3.0 percent sulfur coal and 2 ppm  $\text{NH}_3$  Slip – ABS deposits should form in air heater
- SNCR appears to have deposits at a lower Radian number, probably due to uneven distribution of  $\text{NH}_3$  slip
- Localized area of high  $\text{NH}_3$  slip may form a ring of ABS deposits in air heater, as baskets rotate past the high  $\text{NH}_3$  slip

# Ammonia in Effluent Discharge

## ■ General

- $\text{NH}_3$  is ad/absorbed on fly ash and in FGD slurry
- $\text{NH}_3$  is soluble
- $\text{NH}_3$  eventually ends up in effluent discharge

## ■ Air heater wash water

- Worst case - for 2 ppm  $\text{NH}_3$  slip and all  $\text{NH}_3$  in air heater, ~6,000 ppm  $\text{NH}_3$  in wash water
- Early operation – for 0.5 ppm  $\text{NH}_3$  slip and 20 percent  $\text{NH}_3$  in air heater, ~125 ppm  $\text{NH}_3$  wash water

## ■ Fly ash sluice pond discharge

- Worst case – for 2 ppm  $\text{NH}_3$  slip and 100 percent on fly ash, ~1.5 to 2.0 ppm  $\text{NH}_3$  in fly ash sluice water
- Early operation – for 0.5 ppm  $\text{NH}_3$  slip and 70 percent on fly ash, <0.5 ppm  $\text{NH}_3$

# Ammonia in Effluent Discharge (Cont' d)

## ■ FGD blowdown

- Worst case – for 2 ppm  $\text{NH}_3$  slip and 100 percent in FGD slurry, ~10 ppm  $\text{NH}_3$  in FGD slurry (open loop operation)
- Early operation – for 0.5 ppm  $\text{NH}_3$  slip and 10 percent in FGD slurry, ~1 ppm  $\text{NH}_3$

## ■ Typical application – for 2 ppm $\text{NH}_3$ slip

- Air heater – 20 percent and 1200 ppm
- Fly ash – 70 percent and ~1 ppm
- FGD – 10 percent and ~1 ppm
- Composite – 1.2 ppm (average annual basis)

# Ammonia in Effluent Discharge (Cont' d)

## ■ Summary

- Air heater wash water will have a “slug” of  $\text{NH}_3$  to chemical treatment pond
- Fly ash pond effluent discharge with air heater wash water blending (for each ppm of  $\text{NH}_3$  slip):

5 days	7 ppm $\text{NH}_3$
10 days	3.5 ppm $\text{NH}_3$
30 days	1.5 ppm $\text{NH}_3$

# SO<sub>3</sub> Mitigation

- SO<sub>3</sub> removal location
  - In-furnace – >2500°F and ~2000°F
  - In-ductwork – ~650°F and 300°F
- Alkali for reaction
  - Limestone – either calcitic or dolomitic
  - Quick and hydrated lime - slurry or dry
  - Magnesium hydroxide – commercial and byproduct from Thiosorbic lime FGD
- In-furnace
  - >2500°F – dead burning, furnace slagging and high stoich
  - ~2000°F – using Mg(OH)<sub>2</sub>
- In-ductwork
  - ~650°F – no published data
  - ~300°F – EPRI's High Sulfur Test Center

## S03 Mitigation (cont' d)

- Each 1.0 alkali stoichiometry is approximately equal to 1% of the coal's ash (for high sulfur coals)

# Balancing Ammonia Slip and SO<sub>3</sub>

## ■ Air heater

- Optimize air heater cleaning cycle for ABS and fly ash deposits and for pressure drop – function of SO<sub>3</sub> at air heater inlet

## ■ NH<sub>3</sub> in effluent discharge

- Minimize steady-state NH<sub>3</sub> in effluent discharge: Optimum air heater deposits and treatment of air heater wash water – may eliminate need for SO<sub>3</sub> removal
- If no limit on steady-state NH<sub>3</sub> in effluent discharge: No air heater ABS deposits, where SO<sub>3</sub> removal should be required

# Conclusions

- SO<sub>3</sub> removal system after SCR & before the air heater will:
  - Eliminate the potential for a visible SO<sub>3</sub> plume and eliminate corrosion from higher acid dew point
  - Allow for balancing the NH<sub>3</sub> deposition between the air heater and the ESP to minimize impact on plant operation



# Recommendations

## ■ SO<sub>3</sub> measurements

- Before and during SCR operation – determine where SO<sub>3</sub> is formed and removed
- During SCR operation – determine whether percent removal or ppm removed is constant with increased SO<sub>3</sub>

## ■ NH<sub>3</sub> measurements

- Wash air heater after first ozone season and collect wash water samples to estimate NH<sub>3</sub> in air heater. Monitor NH<sub>3</sub> slip (at SCR outlet) and NH<sub>3</sub> in fly ash → determines NH<sub>3</sub> slip distribution
- After first ozone season, increase NO<sub>x</sub> removal to increase NH<sub>3</sub> slip – to point where deposits are formed in air heater to determine plant specific Radian number